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Pure Substances and Mixtures, Elements \u0026amp;

Compounds, Classification of Matter, Chemistry Examples,

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3.2.1/3.2.2 Describe the differences between elements, compounds and mixtures. ~~States of Matter : Solid Liquid Gas~~  
~~10 Amazing Experiments with Water~~ Changes in the State of Matter - Science Experiments - Solid to Liquid, Liquid to Gas, Gas to Liquid Changes of States of Matter Conservation of Mass in Chemical Reactions The States of Matter - Changes of State - Solid Liquid Gas Plasma | Physical Science | Gen Chemistry States of Matter Understanding Atoms, elements, and molecules Part #1 (9min) Study Jams - Properties of Matter States of Matter and Changes of State - Science for Kids

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What Is Matter? - The Dr. Binocs Show | Best Learning

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145 CHAPTER 9 SOLUTIONS MANUAL 35. Aqueous solutions of potassium iodide and silver nitrate are mixed, forming the precipitate silver iodide. chemical equation:  $KI(aq) + AgNO_3(aq) \rightarrow KNO_3(aq) + AgI(s)$  complete ionic equation:  $K^+(aq) + I^-(aq) + Ag^+(aq) + NO_3^-(aq) \rightarrow K^+(aq) + NO_3^-(aq) + AgI(s)$



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## Chemical Reactions

Solution. Physical: boiling and melting are physical changes.

When water boils, no bonds are broken or formed. The change could be written:  $\text{H}_2\text{O} \left( l \right) \rightarrow \text{H}_2\text{O} \left( g \right)$

Chemical: The dark grey nail changes color to form an orange flaky substance (the rust); this must be a chemical change.

3.6: Changes in Matter: Physical and Chemical Changes ...

240 Chemistry: Matter and Change □ Chapter 12 Solutions Manual CHAPTER 12 SOLUTIONS MANUAL 30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the

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substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

## States of Matter

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